

Post-combustion CO₂ Capture Using Metal Organic Frameworks

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Background

As part of GCEP-sponsored research:

- Develop preliminary systems-level performance and cost models for evaluation of new materials for CO₂ capture
- Incorporate these models in a broader power plant systems model such as the Integrated Environmental Control Model (IECM) to make comparative analyses

Objectives

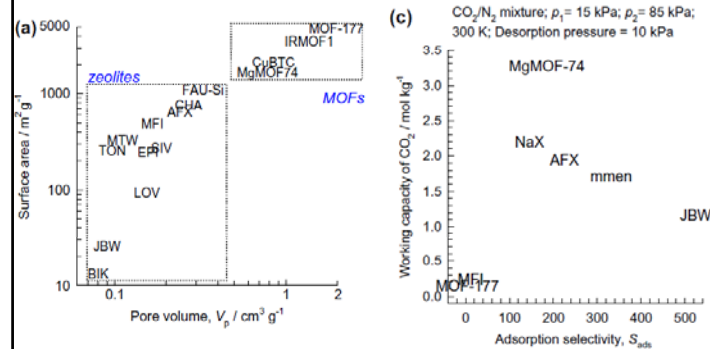
- Develop a preliminary performance model for evaluation of metal organic frameworks (MOFs) for post-combustion CO₂ capture
- Develop a preliminary thermodynamic model for pressure/vacuum swing adsorption (PSA/VSA)
- Compare the performance results with MEA-based CO₂ capture process

Work in progress!

Sorbents for CO₂ capture

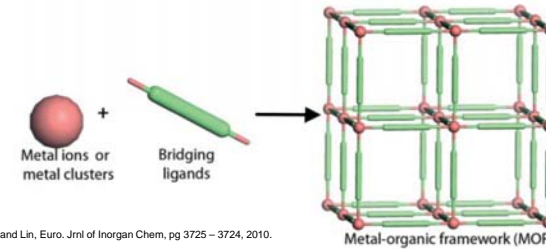
- Zeolites – porous crystalline aluminosilicates
 - Eg. Zeolite13X, NAX
- Amine-functionalized chemisorbents
 - Eg. PEI, NETL-32D
- Metal organic frameworks (MOFs)
 - MOF-5, MOF-177, Mg-MOF-74

MOFs have better CO₂ capture properties than zeolites



Source: Krishna and van Baten, Separation and Purification Technology 87 (2012) 120–126

Metal Organic Frameworks



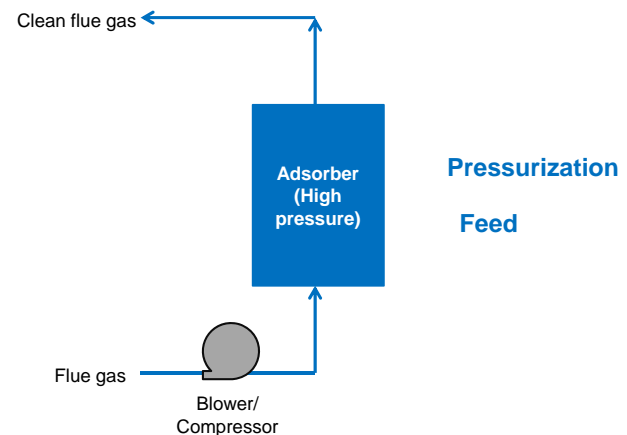
Source: Rocca and Lin, Euro. J. Inorg. Chem., pg 3725 – 3724, 2010.

- Metal-containing nodes linked by organic ligand bridges
- “Tunable” properties to enhance CO₂ capture
- Over 100 MOFs reported in literature
 - Typical metals – Cu, Ni, Al, Sc, Co, Mn
- Pressure swing adsorption and regeneration

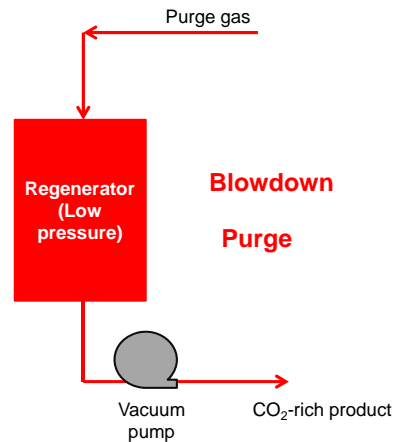
Pressure/vacuum swing adsorption (PSA/VSA)

- Adsorption occurs at high pressure (or at atmospheric pressure in VSA)
- Desorption occurs when pressure is released
- Compared with thermal swing adsorption (TSA)
 - Shorter cycle times
 - Longer sorbent life, but ...
 - Lower CO₂ product purity

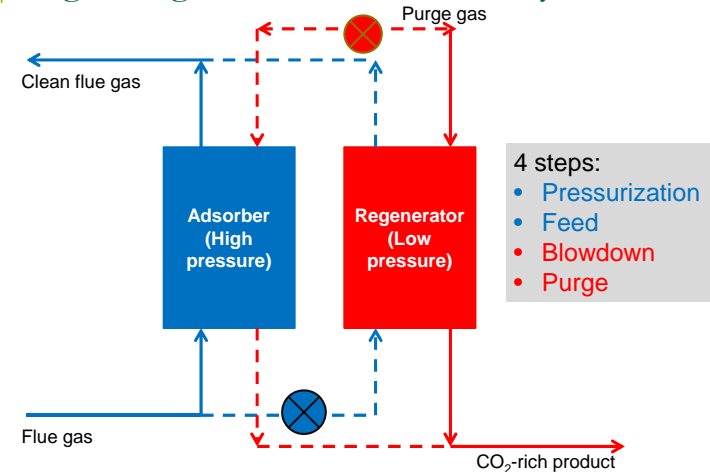
Single-stage PSA – Skarstorm cycle



Single-stage PSA – Skarstorm cycle



Single-stage PSA – Skarstorm cycle



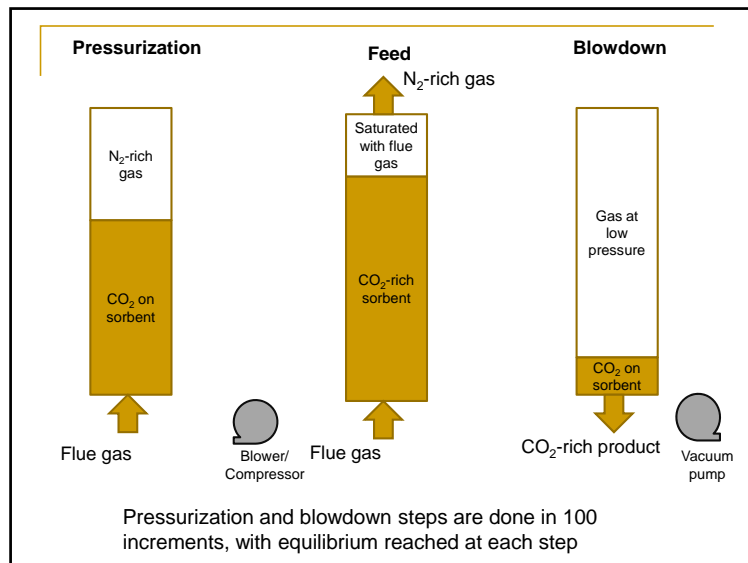
Performance model

- For a given sorbent, desired CO₂ capture efficiency and operating conditions, estimate:
 - Amount of sorbent required
 - Amount of energy required
 - Purity of CO₂ product

Simplified PSA/VSA model*

- Three steps:
 - Pressurization (adsorption)
 - Feed (adsorption)
 - Blowdown (desorption)
- Atmospheric pressure adsorption, vacuum pressure desorption
- Equilibrium conditions
- Cyclic steady state
- Single-stage operation

*Maring and Webley, Intl Jnl of Greenhouse Gas Control, 15, pg 16 – 31, 2013.



A few model equations

$$n_{\text{CO}_2, \text{total}} = n_{\text{CO}_2, \text{adv}} + n_{\text{CO}_2, \text{gas}}$$

$$n_{\text{CO}_2, \text{adv}} = \frac{n_{\text{CO}_2, \text{feed}} n_{\text{CO}_2}}{1 + b_{\text{CO}_2} P_{\text{CO}_2}} \quad b_{\text{CO}_2} = \frac{Q_{\text{CO}_2}}{RT} \quad n_{\text{CO}_2, \text{gas}} = \frac{P_{\text{CO}_2} V}{RT}$$

(Loadings calculated using Langmuir equilibrium model)

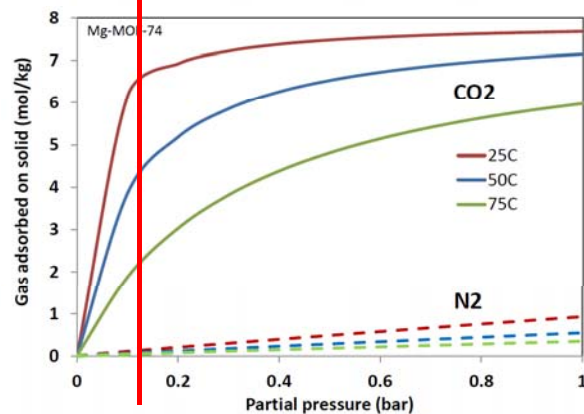
$$n_{\text{CO}_2, \text{product}} = \sum_i n_{\text{CO}_2, i} (n_{\text{total}, i} - n_{\text{total}})$$

$$\text{Purity} = \frac{n_{\text{CO}_2, \text{product}}}{n_{\text{CO}_2, \text{product}} + n_{\text{N}_2, \text{product}}}$$

$$W_{\text{vacuum}} = \sum_i n_{\text{total}, i} (n_{\text{total}, i} - n_{\text{total}}) \times \frac{1}{\eta} \left(\frac{P_i}{P} \right)^{\frac{h-1}{h}} T_{\text{feed}} \left(\left(\frac{P_i}{P} \right)^{\frac{h-1}{h}} - 1 \right)$$

$$\text{Specific work} = \frac{(W_{\text{vacuum}} + W_{\text{blower}})}{n_{\text{CO}_2, \text{product}}}$$

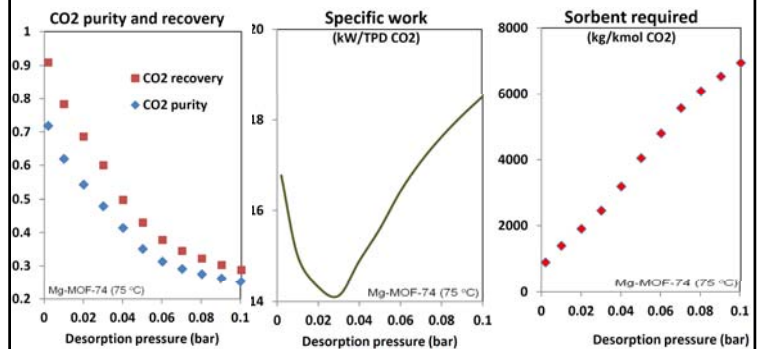
Isotherms for Mg₂(dobdc) – MOF-74



$m_{\text{CO}_2} = 7.9 \text{ mol/kg}$; $b_{0, \text{CO}_2} = 1.56 \times 10^{-6} \text{ /bar}$; $Q_{\text{CO}_2} = 42 \text{ kJ/mol}$
 $m_{\text{N}_2} = 14 \text{ mol/kg}$; $b_{0, \text{N}_2} = 4.96 \times 10^{-5} \text{ /bar}$; $Q_{\text{N}_2} = 18 \text{ kJ/mol}$

Mosconi et al. Energy and Environmental Science, 4, 3039–3049, 2011

Results from the single-stage VSA model



With single-stage VSA, high recovery and purity is possible only at very low desorption pressure.

Preliminary case study

- Base plant (modeled using IECM 8.0.2)
 - 650 MWg, Appalachian Medium Sulfur coal
 - 11,310 kmol/hr CO₂ in flue gas (12% by volume)
- CO₂ capture using Mg-MOF-74 and VSA
 - 90% CO₂ capture
 - Desorption pressure 0.002 bar
 - Isothermal at 75°C
 - CO₂ product compressed to 135 bar

Case study results

	Base plant*	MOF-VSA CO ₂ capture
Gross power out (MW)	650	650
Thermal energy input (MWth)	1564	1564
Net power out (MW)	608	362
Net plant efficiency (%HHV)	39	23

Using a single-stage VSA process, energy penalty using MOFs is much higher compared to conventional MEA-based CO₂ capture

Future work

- Improve the performance model
- Expand the model to incorporate multi-stage and advanced VSA cycles
- Explore better MOF materials
- Explore possibilities of combined PSA-VSA or TSA-VSA cycles
- Develop a cost model

Acknowledgement

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Thank you!

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